## Cambridge International Examinations

## COMBINED SCIENCE

## Additional Materials: Multiple Choice Answer Sheet

Soft clean eraser
Soft pencil (type B or HB is recommended)

## READ THESE INSTRUCTIONS FIRST

Write in soft pencil.
Do not use staples, paper clips, glue or correction fluid.
Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you.
DO NOT WRITE IN ANY BARCODES.

There are forty questions on this paper. Answer all questions. For each question there are four possible answers A, B, C and D.
Choose the one you consider correct and record your choice in soft pencil on the separate Answer Sheet.

## Read the instructions on the Answer Sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer.
Any rough working should be done in this booklet.
A copy of the Periodic Table is printed on page 16.
Electronic calculators may be used.

1 Cell P and cell Q are animal cells.
Cell $P$ is placed into a concentrated salt solution.
Cell Q is placed into distilled (pure) water.
What is the appearance of the two cells after ten minutes?

|  | cell P | cell Q |
| :---: | :---: | :---: |
| A | burst | burst |
| B | burst | shrivelled |
| C | shrivelled | burst |
| D | shrivelled | shrivelled |

2 Every seed contains a food reserve.
What makes this food available to the young plant at germination?
A cellulose
B chlorophyll
C enzymes
D lactic acid

3 The diagram shows an investigation into the effect of light intensity on the rate of photosynthesis. The rate is measured by counting the number of bubbles released per minute.

The experiment is repeated using different light intensities.


Which graph shows the result of the investigation?


increasing light intensity
D


4 Which statement describes the effect of a lack of nitrogen on plants?
A faster root growth and purple leaves
B faster root growth and yellow leaves
C poor plant growth and purple leaves
D poor plant growth and yellow leaves

5 When a child sucks a sweet it may stay in their mouth for some time.
How does this contribute to tooth decay?
A The sugar in the sweet stops bacteria from growing.
B The teeth are damaged by acid being produced in the mouth.
C The teeth are damaged by alkali being produced in the mouth.
D The teeth are damaged by artificial flavourings in the sweet.

6 Which row does not correctly link a component of blood and its function?

|  | component of blood | function |
| :---: | :---: | :---: |
| A | plasma | transports urea |
| B | platelets | blood clotting |
| C | red blood cells | transport oxygen |
| D | white blood cells | transport $\mathrm{CO}_{2}$ |

7 The diagram shows a section through an alveolus and a blood capillary.


Which row describes the oxygen concentrations at $\mathrm{X}, \mathrm{Y}$ and Z ?

|  | X | Y | Z |
| :---: | :---: | :---: | :---: |
| A | high | low | high |
| B | high | low | low |
| C | low | high | high |
| D | low | high | low |

8 What is the excretory product in blood that is removed by the lungs?
A carbon dioxide
B glucose
C lactic acid
D urea

9 Which processes take place in the eye when a person moves into dim light?

|  | size of pupil | circular muscles <br> of iris | radial muscles <br> of iris |
| :---: | :---: | :---: | :---: |
| A | enlarges | contract | relax |
| B | enlarges | relax | contract |
| C | reduces | contract | relax |
| D | reduces | relax | contract |

10 What is not an effect of excessive alcohol consumption?
A addiction
B depression
C lack of self control
D quicker reaction times

11 What is the main source of energy for food webs?
A chemical reactions
B heat from the centre of the Earth
C heat from the Sun
D light from the Sun

12 Which are results of deforestation in tropical rainforests?
A increased atmospheric carbon dioxide and decreased biodiversity
B increased atmospheric carbon dioxide and increased biodiversity
C increased atmospheric oxygen and decreased biodiversity
D increased atmospheric oxygen and increased biodiversity

13 Which combination of factors is least likely to stop menstruation?

|  | diet | stress |
| :---: | :---: | :---: |
| A | balanced | high |
| B | balanced | low |
| C | unbalanced | high |
| D | unbalanced | low |

14 Powdered magnesium carbonate is slowly added to dilute sulfuric acid.
Which pieces of apparatus are needed to continue the experiment to obtain a sample of magnesium sulfate crystals?


1


2


3
A 1 and 3
B 1 only
C 2 and 3
D 3 only

15 An atom of sodium is represented by ${ }_{11}^{23} \mathrm{Na}$.
What are the numbers of neutrons and protons in this atom?

|  | number of <br> neutrons | number of <br> protons |
| :---: | :---: | :---: |
| A | 11 | 12 |
| B | 12 | 11 |
| C | 23 | 11 |
| D | 23 | 12 |

16 Which atom forms an ion with a charge of 2+?

|  | proton (atomic) number |
| :---: | :---: |
| A | 6 |
| B | 8 |
| C | 12 |
| D | 16 |

17 The table shows the electronic structure of four elements.

| element | electronic structure |
| :---: | :---: |
| W | 2,6 |
| X | 2,8 |
| Y | $2,8,1$ |
| Z | $2,8,7$ |

Which pair of atoms form a covalent molecule?
A two atoms of W
B two atoms of $X$
C an atom of $W$ and an atom of $X$
D an atom of Y and an atom of Z

18 Which formula has the greatest number of atoms?
A $\mathrm{Ca}_{3}\left(\mathrm{PO}_{4}\right)_{2}$
B $\mathrm{Cu}\left(\mathrm{CH}_{3} \mathrm{COO}\right)_{2}$
C $\mathrm{Fe}_{2}\left(\mathrm{SO}_{4}\right)_{3}$
D $\left(\mathrm{NH}_{4}\right)_{2} \mathrm{CO}_{3}$

19 An element forms an amphoteric oxide.
Which substances will react with this amphoteric oxide?

|  | acids | alkalis |  |
| :---: | :---: | :---: | :---: |
| A | $\checkmark$ | $\checkmark$ | key |
| B | $\checkmark$ | $x$ | $\checkmark$ = reacts |
| C | $x$ | $\checkmark$ | $x=$ does not react |
| D | $x$ | $x$ |  |

20 An element $X$ from Period 2 in the Periodic Table is heated in air.
It forms an oxide which dissolves in water.
Universal Indicator added to the solution turns blue.
Which row describes element $X$ ?

|  | metal or non-metal | position in the period |
| :---: | :---: | :---: |
| A | metal | on the left side |
| B | metal | on the right side |
| C | non-metal | on the left side |
| D | non-metal | on the right side |

21 Which diagram represents the structure of an alloy?

A


B


C


D


22 Part of the reactivity series of metals is given below.
barium
calcium
magnesium
chromium
iron
copper
palladium
platinum

Which statement is not correct?
A Barium will not react with cold water but will react with steam.
B Chromium will react with dilute hydrochloric acid.
C Palladium will not react with cold water or steam.
D Platinum will not react with dilute hydrochloric acid.

23 Which two gases are both pollutants of the atmosphere?
A carbon monoxide and oxygen
B carbon monoxide and sulfur dioxide
C nitrogen and oxygen
D nitrogen and sulfur dioxide

24 Ammonia is made in the Haber process.
Which conditions are used in the Haber process?

|  | temperature <br> $/{ }^{\circ} \mathrm{C}$ | pressure <br> $/ \mathrm{atm}$ | catalyst |
| :---: | :---: | :---: | :---: |
| A | 200 | 450 | aluminium oxide |
| B | 200 | 450 | iron |
| C | 450 | 200 | aluminium oxide |
| D | 450 | 200 | iron |

25 The diagrams show the structures of four organic molecules.
P

Q
R

S


Which two are members of the same homologous series?
A Pand R
B Pand S
C $Q$ and $R$
D R and S

26 Which statement about alkenes is not correct?
A Alkenes are made by cracking.
B Alkenes are saturated compounds.
C Alkenes contain carbon-carbon double bonds.
D Alkenes turn bromine water colourless.

27 The properties of ethanol make it a very useful substance.
Which statement is not correct?
A Ethanol burns to form carbon dioxide and water and is often used as a fuel.
B Ethanol can be formed from sugar and is found in wine.
C Ethanol is a useful solvent because it is able to dissolve many different substances.
D Ethanol is very slippery and so is used as a lubricant.

28 The diagram shows a method of measuring the diameter of a beaker.


What is the diameter of the beaker?
A 4.5 cm
B 5.0 cm
C 5.5 cm
D 8.0 cm

29 The gradient of the line on a graph gives the acceleration of a moving object.
What are the quantities on the horizontal and vertical axes of this graph?

|  | quantity on <br> horizontal axis | quantity on <br> vertical axis |
| :---: | :---: | :---: |
| A | speed | distance |
| B | speed | time |
| C | time | distance |
| D | time | speed |

30 An unbalanced force is applied to a moving object.
Which statement does not describe the possible motion of the object?
A the object accelerates
B the object's direction alters
C the object's speed decreases
D the object's velocity remains constant

31 The diagrams show how a spring extends when a weight of 6.0 N is hung on it.


Which weight hanging from the spring causes the length to become 15 cm ?
A 7.5 N
B 15 N
C 30 N
D 45 N

32 Which type of energy is converted to thermal energy when atoms combine?
A chemical
B kinetic
C nuclear
D solar

33 A beaker of water contains a red crystal which is slowly dissolving.
Gentle heat is applied below the crystal.


The red colour rises.
What is the name of this process?
A conduction
B convection
C evaporation
D radiation

34 Which line in the table shows examples of transverse and longitudinal waves?

|  | transverse | longitudinal |
| :---: | :---: | :---: |
| A | gamma-rays | light |
| B | infra-red | sound |
| C | radio | water waves |
| D | sound | X-rays |

35 The ray diagram shows light reflecting off a plane mirror.


What is the angle between the incident and reflected rays?
A $40^{\circ}$
B $50^{\circ}$
C $80^{\circ}$
D $100^{\circ}$

36 The diagram shows four charged objects, $P, Q, R$ and $S$.

positively charged

negatively charged

unknown charge

unknown charge

It is found that $P$ attracts $R$ but repels $S$.
Which statement is correct?
A $Q$ attracts $R$.
B Q repels $S$.
C $R$ attracts $S$.
D R repels $S$.

37 The circuit shown is used to determine the resistance of a lamp for two different brightness settings.


When the lamp brightness is low, the voltmeter reading is 2 V and the ammeter reading is 2 A .
When the lamp brightness is normal, the readings are 12 V and 4 A .
What was the increase in filament resistance?
A $1 \Omega$
B $2 \Omega$
C $3 \Omega$
D $4 \Omega$

38 A 750 W microwave oven is used in a house where the mains voltage is 240 V .
Which fuse should be used in the plug?
A 3A
B 5 A
C $\quad 10 \mathrm{~A}$
D 13 A

39 The diagram shows a transformer.


There is a current in the primary coil and another current in the secondary coil.
Why is there a current in the secondary coil?
A because electricity is conducted through the core
B because the current in the primary coil does not change
C because there are more turns on the primary coil than on the secondary coil
D because there is a changing magnetic field in the core

40 The count rate produced by a source containing a radioactive nuclide falls from 1200 to 75 in 3 minutes.

What is the half-life of the radioactive nuclide?
A 0.75 minutes
B 1.0 minutes
C 3.0 minutes
D 12.0 minutes

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The volume of one mole of any gas is $24 \mathrm{dm}^{3}$ at room temperature and pressure (r.t.p.).

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